DuPont™ Suva®
refrigerants

DuPont
HCFC-124
Properties, Uses, Storage, and Handling

DuPont™ Suva® refrigerants
DuPont fire extinguishant
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Introduction

Background
Chlorofluorocarbons (CFCs), which were developed over 60 years ago, have many unique properties. They are low in toxicity, nonflammable, noncorrosive, and compatible with other materials. In addition, they offer the thermodynamic and physical properties that make them ideal for a variety of uses. CFCs have been used as refrigerants; as blowing agents in the manufacture of insulation, packaging, and cushioning foams; as cleaning agents for metal and electronic components; and in many other applications.

Hydrochlorofluorocarbon 124
HCFC-124 is a component of blends intended to replace R-12 in refrigeration applications. It may also be used as a pure refrigerant to replace R-114 in limited specific applications. HCFC-124 also has potential utility as a fire extinguishant. It may be identified by any of the following names:

- Hydrochlorofluorocarbon 124
- 2-chloro-1,1,1,2-tetrafluoroethane
- HCFC-124
- Suva® 124 refrigerant
- FE-241 fire extinguishant

Table 1 gives the chemical name and formula for HCFC-124.

Uses
Refrigeration
HCFC-124 is a component of DuPont™ Suva® MP39 and Suva® MP66 refrigerants. These refrigerants closely match the performance of R-12 and, along with Suva® 134a, are DuPont’s primary replacements for R-12 and R-500 in a wide range of applications. R-12 and R-500 have been used in supermarkets, air conditioning, food service, and transport refrigeration. Suva® MP refrigerants offer equivalent or improved capacity and energy efficiency in these applications.

HCFC-124 may be used by itself as a pure refrigerant to replace R-114 in some applications, although its vapor pressure is somewhat higher than that of R-114. In addition, when replacing R-114 with HCFC-124, significantly higher refrigeration capacity will be encountered. This may result in electrical overload of the motor unless precautions are taken. The expansion device may also need to be modified.

Fire Suppression
DuPont is actively pursuing HCFC-124 as a transitional product to replace Halon 1301 and Halon 1211 in total flooding and streaming applications.

<table>
<thead>
<tr>
<th>Chemical Name</th>
<th>Formula</th>
<th>CAS Number</th>
<th>Molecular Weight</th>
</tr>
</thead>
<tbody>
<tr>
<td>HCFC-124</td>
<td>2-chloro-1,1,1,2-tetrafluoroethane</td>
<td>2837-89-0</td>
<td>136.48</td>
</tr>
</tbody>
</table>

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Physical Properties

Physical properties of HCFC-124 are shown in Table 2.

Pressure-enthalpy diagrams for HCFC-124 are shown in Figures 1 and 2.

Thermodynamic tables in English and SI units are available in Bulletins T-124-ENG (H-47755) and T-124-SI (H-47756).

Chemical/Thermal Stability

Stability with Metals

Stability tests for refrigerant with metals are typically performed in the presence of refrigeration lubricants. Results of sealed tube stability tests are available for R-12/mineral oil and alkylbenzene lubricants, which have shown long-term stability in contact with copper, steel, and aluminum in actual refrigeration systems. Mineral oils and alkylbenzenes are possible candidates for use with refrigerants containing HCFC-124. Tests with HCFC-based refrigerants in general show acceptable compatibility and stability with alkylbenzene lubricants. As can be seen in Tables 3 and 4, HCFC-124 is significantly more stable than CFC-114. This is indicated by the much lower chloride generation in the stability tests.

Note: Lubricant/refrigerant combinations shown throughout this report are for the purposes of comparing the stability and compatibility of different lubricants with HCFC-124. No recommendation is made or implied that these combinations will operate successfully in refrigeration systems.

<table>
<thead>
<tr>
<th>Physical Property</th>
<th>Units</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Molecular Weight (avg.)</td>
<td>kg/m³</td>
<td>136.48</td>
</tr>
<tr>
<td>Boiling Point at 1 atm</td>
<td>°C</td>
<td>-12.1</td>
</tr>
<tr>
<td></td>
<td>°F</td>
<td>10.3</td>
</tr>
<tr>
<td>Critical Temperature</td>
<td>°C</td>
<td>122.2</td>
</tr>
<tr>
<td></td>
<td>°F</td>
<td>252.0</td>
</tr>
<tr>
<td>Critical Pressure</td>
<td>kPa</td>
<td>3614</td>
</tr>
<tr>
<td></td>
<td>psia</td>
<td>524.5</td>
</tr>
<tr>
<td>Critical Density</td>
<td>kg/m³</td>
<td>565</td>
</tr>
<tr>
<td></td>
<td>lb/ft³</td>
<td>35.28</td>
</tr>
<tr>
<td>Liquid Density at 25°C (77°F)</td>
<td>kg/m³</td>
<td>1401</td>
</tr>
<tr>
<td></td>
<td>lb/ft³</td>
<td>87.6</td>
</tr>
<tr>
<td>Density, Satd. Vapor at -15°C (5°F)</td>
<td>kg/m³</td>
<td>6.88</td>
</tr>
<tr>
<td></td>
<td>lb/ft³</td>
<td>0.43</td>
</tr>
<tr>
<td>Specific Heat Liquid at 25°C</td>
<td>kJ/kg•°C</td>
<td>1.13</td>
</tr>
<tr>
<td></td>
<td>Btu/lb•°F</td>
<td>0.27</td>
</tr>
<tr>
<td>Specific Heat Vapor at 25°C and 1 atm</td>
<td>kJ/kg•°C</td>
<td>0.742</td>
</tr>
<tr>
<td></td>
<td>Btu/lb•°F</td>
<td>0.177</td>
</tr>
<tr>
<td>Vapor Pressure at 25°C</td>
<td>kPa</td>
<td>386</td>
</tr>
<tr>
<td></td>
<td>psia</td>
<td>56</td>
</tr>
<tr>
<td>Heat of Vaporization at Boiling Point</td>
<td>kJ/kg</td>
<td>194</td>
</tr>
<tr>
<td></td>
<td>Btu/lb</td>
<td>83.2</td>
</tr>
<tr>
<td>Thermal Conductivity at 25°C</td>
<td>W/m•K</td>
<td>0.0722</td>
</tr>
<tr>
<td></td>
<td>Btu/hr•ft•°F</td>
<td>0.0417</td>
</tr>
<tr>
<td></td>
<td>W/m•K</td>
<td>0.0106</td>
</tr>
<tr>
<td></td>
<td>Btu/hr•ft•°F</td>
<td>0.0061</td>
</tr>
<tr>
<td>Viscosity at 25°C</td>
<td>mPa•s</td>
<td>0.299</td>
</tr>
<tr>
<td></td>
<td>cP</td>
<td>0.299</td>
</tr>
<tr>
<td></td>
<td>mPa•s</td>
<td>0.0120</td>
</tr>
<tr>
<td></td>
<td>cP</td>
<td>0.0120</td>
</tr>
<tr>
<td>Flammability Limit in Air at 1 atm</td>
<td>vol%</td>
<td>None</td>
</tr>
<tr>
<td>Ozone Depletion Potential</td>
<td>(CFC-12 = 1)</td>
<td>0.02</td>
</tr>
<tr>
<td>Halocarbon Global Warming Potential</td>
<td>(CFC-11 = 1)</td>
<td>0.10</td>
</tr>
<tr>
<td>TSCA Inventory Status</td>
<td>Reported/Included?</td>
<td>Yes, consent order</td>
</tr>
<tr>
<td>Inhalation Exposure Limit</td>
<td>AEL* (8 and 12 hr TWA)</td>
<td>1000</td>
</tr>
</tbody>
</table>

*Acceptable Exposure Limit (AEL) is an airborne inhalation exposure limit established by DuPont that specifies time-weighted average concentrations to which nearly all workers may be repeatedly exposed without adverse effects.
Figure 1. Pressure-Enthalpy Diagram for HCFC-124 (English Units)
Figure 2. Pressure-Enthalpy Diagram for HCFC-124 (SI Units)
Table 3
Compatibility with Metal Compressor Materials

<table>
<thead>
<tr>
<th>Material</th>
<th>HCFC-124 (Alkylbenzene)</th>
<th>CFC-114 (Paraffinic Mineral Oil)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Aluminum Impeller</td>
<td>3.7</td>
<td>45.3</td>
</tr>
<tr>
<td>Copper Tube</td>
<td>0.2</td>
<td>30.3</td>
</tr>
<tr>
<td>Titanium Tube</td>
<td>2.1</td>
<td>53.7</td>
</tr>
<tr>
<td>Gray Cast Iron</td>
<td>0.4</td>
<td>43.0</td>
</tr>
<tr>
<td>Steel Tubing</td>
<td>0.8</td>
<td>12.1</td>
</tr>
</tbody>
</table>

Aged 14 days at 175°C

Table 4
Chemical Stability

<table>
<thead>
<tr>
<th>Lubricant</th>
<th>Average Chloride Ion Values, µg</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>HCFC-124</td>
</tr>
<tr>
<td>Naphthenic</td>
<td>120</td>
</tr>
<tr>
<td>Alkylbenzene</td>
<td>105</td>
</tr>
<tr>
<td>Paraffinic</td>
<td>166</td>
</tr>
</tbody>
</table>

Aged 14 days at 175°C

Copper, steel, and aluminum coupons

Thermal Decomposition
Like R-12 and R-114, HCFC-124 will decompose when exposed to high temperatures from sources such as open flames or electric resistance heaters. Decomposition may produce toxic and irritating compounds such as hydrochloric and hydrofluoric acid and possibly carbonyl halides. The pungent odors released will irritate the nose and throat and generally force people to evacuate the area. Therefore, it is important to prevent decomposition by following DuPont recommendations for handling and use.

Concerns if R-12 or R-114 and Refrigerants Containing HCFC-124 Are Mixed
R-12 and R-114 and HCFC-124 are chemically compatible with each other; this means that they do not react with each other and form other compounds. However, when different refrigerants are mixed by accident or deliberately, they will form mixtures that are very difficult to separate. Therefore, mixtures of R-12 or R-114 and HCFC-124 containing refrigerants cannot be separated in on-site recycle machines or in the typical facilities of an off-site reclamer. These mixtures will have to be disposed of by incineration.

Also, mixtures of R-12 or R-114 and HCFC-124 containing refrigerants will have performance properties different than either refrigerant alone. These properties may not be acceptable for your systems. We do not recommend mixing refrigerants in any system. First remove the R-12 or R-114 properly and then charge the new refrigerant.
Materials Compatibility
Because HCFC-124 will be used in different applications, it is important to review materials of construction for compatibility when designing new equipment, retrofitting existing equipment, or preparing storage and handling facilities. The compatibility data summarized here will include materials commonly used in refrigeration applications.

Elastomers and Plastics
It should be recognized that these data reflect compatibility in sealed tube tests, and that HCFC-124 compatibility in real systems can be influenced by the actual operating conditions, the nature of the polymers used, compounding formulations of the polymers, and the curing or vulcanization processes used to create the polymer. Elastomers should always be tested under actual operating conditions before reaching final conclusions about their suitability.

The rankings shown in Table 5 are based on samples of each polymer subjected to aging at room temperature for two weeks in HCFC-124 alone. Table 6 shows compatibility of HCFC-124 with various plastics using the same aging conditions. Physical properties of the test samples were determined before and after aging. The resulting ratings are based on zero (0) being best and two (2) being worst for the purposes of comparison. The factors included in the overall assessment of compatibility included:

• visual observations of material changes due to aging
• changes in weight and volume of the samples due to aging
• changes in hardness of the samples due to aging
• changes in flexural properties of the samples due to aging

Desiccants
In refrigeration systems, keeping the refrigerant and lubricant free of moisture is very important. Driers filled with moisture-adsorbing desiccant are typically used to prevent moisture accumulation.

Pure HCFC-124 is reported by UOP to be compatible with their XH-6 and XH-9 molecular sieve desiccants and by W. R. Grace to be compatible with their MS 592 or MS 594 molecular sieve desiccants. Equivalent molecular sieves produced by other manufacturers may also be used. Loose filled and compacted bead driers contain only molecular sieve desiccants, while solid core driers contain molecular sieves mixed with other materials. Consult the manufacturers of solid core driers for their recommendations.

Suva® refrigerant blends containing HCFC-124 plus other components may not be compatible with XH-6 molecular sieve desiccant. Refer to the appropriate DuPont bulletins for information on these refrigerants.

Refrigeration Lubricants
Most compressors require a lubricant to protect internal moving parts. The compressor manufacturer usually recommends the type of lubricant(s) and proper viscosity that should be used to ensure acceptable operation and equipment durability. Recommendations are based on several criteria, which can include lubricity, miscibility, compatibility with materials of construction, thermal stability, and compatibility with other lubricants. It is important to follow the manufacturers’ recommendations for lubricants to be used with their equipment.
<table>
<thead>
<tr>
<th>Polymer</th>
<th>Rating</th>
<th>Linear Swell, %</th>
<th>Weight Gain, %</th>
<th>Hardness Change</th>
</tr>
</thead>
<tbody>
<tr>
<td>Natural Rubber</td>
<td>0</td>
<td>3</td>
<td>22</td>
<td>−7</td>
</tr>
<tr>
<td>Butyl Rubber</td>
<td>0</td>
<td>&lt;1</td>
<td>8</td>
<td>−8</td>
</tr>
<tr>
<td>Nords EPDM</td>
<td>0</td>
<td>−1</td>
<td>3</td>
<td>−5</td>
</tr>
<tr>
<td>Neoprene CR</td>
<td>0</td>
<td>&lt;1</td>
<td>3</td>
<td>−4</td>
</tr>
<tr>
<td>SBR</td>
<td>1</td>
<td>&lt;1</td>
<td>6</td>
<td>−15</td>
</tr>
<tr>
<td>Nitrile Rubber</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>NBR</td>
<td>0</td>
<td>&lt;1</td>
<td>10</td>
<td>−7</td>
</tr>
<tr>
<td>HNBR</td>
<td>2</td>
<td>28</td>
<td>169</td>
<td>−17</td>
</tr>
<tr>
<td>Vamac EA</td>
<td>2*</td>
<td>32</td>
<td>175</td>
<td>−14</td>
</tr>
<tr>
<td>Hypalon CSM</td>
<td>0</td>
<td>3</td>
<td>9</td>
<td>−5</td>
</tr>
<tr>
<td>NBR</td>
<td>0</td>
<td>&lt;1</td>
<td>10</td>
<td>−7</td>
</tr>
<tr>
<td>HNBR</td>
<td>2</td>
<td>28</td>
<td>169</td>
<td>−17</td>
</tr>
<tr>
<td>Vamac EA</td>
<td>2*</td>
<td>32</td>
<td>175</td>
<td>−14</td>
</tr>
<tr>
<td>Silicone</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Viton A</td>
<td>2</td>
<td>16</td>
<td>57</td>
<td>−13</td>
</tr>
<tr>
<td>Viton B</td>
<td>2</td>
<td>23</td>
<td>85</td>
<td>−34</td>
</tr>
<tr>
<td>Viton GF</td>
<td>2</td>
<td>25</td>
<td>88</td>
<td>−25</td>
</tr>
<tr>
<td>Kalrez</td>
<td>2</td>
<td>19</td>
<td>47</td>
<td>—</td>
</tr>
<tr>
<td>Fluorinated Silicone</td>
<td>2</td>
<td>17</td>
<td>81</td>
<td>−16</td>
</tr>
<tr>
<td>Silicone</td>
<td>2</td>
<td>19</td>
<td>125</td>
<td>−12</td>
</tr>
<tr>
<td>Epichlorohydrin</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Homopolymer</td>
<td>0</td>
<td>−0.2</td>
<td>6</td>
<td>−4</td>
</tr>
<tr>
<td>Adiprene U</td>
<td>2</td>
<td>29</td>
<td>172</td>
<td>−18</td>
</tr>
<tr>
<td>FA Polysulfide</td>
<td>0</td>
<td>−1</td>
<td>3</td>
<td>−2</td>
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<tr>
<td>Thermoplastics</td>
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<td></td>
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<td></td>
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<tr>
<td>Alcryr</td>
<td>0</td>
<td>7</td>
<td>38</td>
<td>8</td>
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<tr>
<td>Santoprene</td>
<td>0</td>
<td>1</td>
<td>12</td>
<td>−7</td>
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<tr>
<td>Geolast</td>
<td>1</td>
<td>8</td>
<td>70</td>
<td>−13</td>
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<tr>
<td>Hytrel polyester</td>
<td>0</td>
<td>8</td>
<td>36</td>
<td>0</td>
</tr>
<tr>
<td>Gasketing Materials</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cork</td>
<td>—</td>
<td>2</td>
<td>134</td>
<td>—</td>
</tr>
<tr>
<td>Vegetable fiber</td>
<td>—</td>
<td>−0.2</td>
<td>34</td>
<td>—</td>
</tr>
<tr>
<td>Gylon filled PTFE</td>
<td>—</td>
<td>2</td>
<td>6</td>
<td>—</td>
</tr>
<tr>
<td>Armalon PTFE felt</td>
<td>—</td>
<td>2</td>
<td>32</td>
<td>—</td>
</tr>
</tbody>
</table>

**RATING:**

“0”—both linear swell and hardness change are less than 10%

“1”—either swell OR hardness change greater than 10%

“2”—both swell AND hardness change greater than 10%

**COMMENTS:**

*sample lost elasticity; did not return to original shape when deformed

**TEST CONDITIONS:**

Two weeks at room temperature; refrigerant only
Current lubricants used with R-12 and R-114 have nearly complete miscibility, which eases the problems of designing systems to allow lubricant return back to the compressor. Many refrigeration systems take advantage of this miscibility when considering lubricant return.

HCFC-124 exhibits less miscibility with common mineral oil lubricants used with R-12 or R-114. However, alkylbenzene appears to be a good lubricant candidate for HCFC-124.

Compressor and equipment manufacturers will recommend lubricants to use with their equipment. It would be difficult to summarize all possible lubricant candidates that may be screened by various equipment manufacturers. In addition, there will be continuing research and development of new lubricants, because the market for alternative refrigerants continues to stimulate other market areas.

Table 7 shows a summary of miscibility tests done with an 80/20 volume mixture of refrigerant and lubricant over a wide range of temperatures, with visual inspection for phase separation as the tubes are slowly warmed. Alkylbenzene offers excellent miscibility with HCFC-124. This table does not show that any refrigerant/lubricant combination is acceptable, only whether the two appear to be miscible at the conditions shown.

### Table 6

**Compatibility of HCFC-124 with Plastics**

<table>
<thead>
<tr>
<th>Plastic</th>
<th>Rating</th>
<th>Weight Change, %</th>
</tr>
</thead>
<tbody>
<tr>
<td>Polypropylene</td>
<td>1</td>
<td>2</td>
</tr>
<tr>
<td>Styron® polystyrene</td>
<td>1</td>
<td>6</td>
</tr>
<tr>
<td>Polyvinyl chloride</td>
<td></td>
<td></td>
</tr>
<tr>
<td>PVC</td>
<td>0</td>
<td>0</td>
</tr>
<tr>
<td>CPVC</td>
<td>0</td>
<td>&lt;1</td>
</tr>
<tr>
<td>Fluorocarbon</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Teflon® PTFE</td>
<td>1</td>
<td>4</td>
</tr>
<tr>
<td>Tefzel® ETFE</td>
<td>2</td>
<td>5</td>
</tr>
<tr>
<td>PVDF</td>
<td>0</td>
<td>&lt;1</td>
</tr>
<tr>
<td>Surlyn® ionomer</td>
<td>1</td>
<td>2</td>
</tr>
<tr>
<td>Lucite® acrylic</td>
<td>2*</td>
<td>—</td>
</tr>
<tr>
<td>Kralastic® ABS</td>
<td>2*</td>
<td>—</td>
</tr>
<tr>
<td>Ethocel® cellulose</td>
<td>2*</td>
<td>40</td>
</tr>
<tr>
<td>Epoxy</td>
<td>0</td>
<td>0</td>
</tr>
<tr>
<td>Delrin® acetal</td>
<td>0</td>
<td>1</td>
</tr>
<tr>
<td>Noryl® PPO</td>
<td>0</td>
<td>&lt;1</td>
</tr>
<tr>
<td>Tuffak® PC</td>
<td>0</td>
<td>&lt;1</td>
</tr>
<tr>
<td>Polyester</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Rynite® PET</td>
<td>0</td>
<td>&lt;1</td>
</tr>
<tr>
<td>Valox® PBT</td>
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<td>&lt;1</td>
</tr>
<tr>
<td>Aylon® polyarylate</td>
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<td>&lt;1</td>
</tr>
<tr>
<td>Xydar® LCP</td>
<td>0</td>
<td>0</td>
</tr>
<tr>
<td>Zytel® nylon</td>
<td>0</td>
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<tr>
<td>Polyimide</td>
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<tr>
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<td>1</td>
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<tr>
<td>Ultem® PEI</td>
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<tr>
<td>Torlan® PAI</td>
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<tr>
<td>Ryton® PPS</td>
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<tr>
<td>Polysulphone</td>
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<tr>
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<td>0</td>
<td>&lt;1</td>
</tr>
<tr>
<td>Radel® Polyzersulfone</td>
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<td>0</td>
</tr>
</tbody>
</table>

**RATING:**

*0*—compatible by visual inspection and weight change

*1*—borderline by visual inspection or weight change

*2*—incompatible by visual inspection and weight change

**COMMENTS:**

*sample dissolved

**TEST CONDITIONS:**

Two weeks at room temperature; refrigerant only
Table 7
Miscibility Summary (Temperature in °C)

<table>
<thead>
<tr>
<th></th>
<th>R-114 with 20% mineral oil</th>
<th>HCFC-124 with 20% mineral oil</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>−60°C −22 +65</td>
<td>−60 +47 +65</td>
</tr>
<tr>
<td></td>
<td>2 phases 1 phase</td>
<td>2 phases 1 phase</td>
</tr>
<tr>
<td></td>
<td>−60 −44 +65</td>
<td></td>
</tr>
<tr>
<td></td>
<td>2 phases 1 phase</td>
<td></td>
</tr>
</tbody>
</table>

Safety


ASHRAE Classification = A1.

Inhalation Toxicity

HCFC-124 poses no acute or chronic hazard when it is handled in accordance with DuPont recommendations and when exposures are maintained below recommended limits, such as the DuPont Acceptable Exposure Limit (AEL) of 1000 ppm, 8 or 12 hour time weighted average (TWA).

An AEL is an airborne exposure limit established by DuPont that specifies time-weighted average airborne concentrations to which nearly all workers may be repeatedly exposed during a 40-hr work week without adverse effects.

Like R-114, exposure above the recommended limit to the vapors of HCFC-124 by inhalation may cause human health effects that can include temporary nervous system depression with anesthetic effects such as dizziness, headache, confusion, loss of coordination, and even loss of consciousness. Higher exposures to the vapors may cause temporary alteration of the heart’s electrical activity with irregular pulse, palpitations, or inadequate circulation. Death can occur from gross overexposure. Intentional misuse or deliberate inhalation of HCFC-124 vapors may cause death without warning. This practice is extremely dangerous.

If you are exposed to HCFC-124 vapors and experience any of the above symptoms, move immediately to fresh air and seek medical attention.

Cardiac Sensitization

An effect that occurs with most hydrocarbons and halocarbons at high concentrations is that the human heart can become sensitized to adrenalin, leading to cardiac irregularities and even cardiac arrest. The likelihood of these cardiac problems increases if you are under physical or emotional stress. HCFC-124 can cause these responses well above the AEL.

If you are exposed to very high concentrations of HCFC-124, move immediately from the area and seek medical attention as a precaution. DO NOT attempt to remain in the area to fix a leak or perform other duties—the effects of over-exposure can be very sudden.

Medical attention must be given immediately if someone is having symptoms of overexposure to HCFC-124. DO NOT treat the patient with drugs such as epinephrine, because these drugs could increase the risk of cardiac problems. If the person is having trouble breathing, administer oxygen. If breathing has stopped, give artificial respiration immediately.
**Skin and Eye Contact**

At room temperature, HCFC-124 vapors have little or no effect on the skin or eyes. However, in liquid form, HCFC-124 can cause frostbite. If contact with liquid does occur, soak the exposed areas in lukewarm water, not cold or hot. In all cases, seek medical attention as soon as possible.

Always wear protective clothing when there is a risk of exposure to liquid refrigerants. Where splashing of refrigerant may occur, always wear eye protection and a face shield.

**Spills or Leaks**

If a large release of vapor occurs, such as from a large spill or leak, the vapors may concentrate near the floor or in low elevation areas, which can displace the oxygen needed for life, resulting in suffocation.

Evacuate everyone until the area has been well ventilated. Re-enter the area only while using self-contained breathing apparatus. Use blowers or fans to circulate the air at floor or low levels.

Always use self-contained breathing apparatus or an air-line respirator when entering tanks or other areas where vapors might exist. Use the buddy system (a second employee stationed outside the tank) and a lifeline. Refer to the Material Safety Data Sheet (MSDS) for HCFC-124 for more complete safety information.

HCFC-124 has virtually no odor and, therefore, can be extremely difficult to detect in enclosed areas. Frequent leak checks and the installation of permanent leak detectors may be necessary for enclosed areas or machinery rooms. Refer to ASHRAE Standards 15 and 34 for machinery room requirements.

To ensure safety when using HCFC-124 refrigerant in enclosed areas:

- Make sure the work area is free of vapors prior to beginning any work.
- Install air monitoring equipment to detect leaks.
- Equipment should *never* be leak tested with a pressurized mixture of air and HCFC-124. A mixture of dry nitrogen and HCFC-124 can be used.

**Storage Tanks**

- Bulk storage tanks should be evacuated prior to initial filling and should *never* be under positive air pressure.
- Tanks should not be pressurized above 1.5 times the normal working pressure with HCFC-124 refrigerant. Relief devices on either the tanks or refrigerant supply system should be verified to be set below this pressure.
- Tank pressures should be monitored routinely to look for accumulation of air.
- Air lines should *never* be connected to refrigerant storage tanks.

**Filling and Charging Operations**

- Before starting work, read the DuPont Material Safety Data Sheet (MSDS) for HCFC-124.
- Before evacuating cylinders or refrigeration equipment, any remaining refrigerant should be removed using a recovery system.
- Vacuum pump discharge lines should be free of restrictions that could increase discharge pressures.
- Cylinders or refrigeration equipment should normally be evacuated at the start of filling and never be under positive air pressure.
- Filled cylinders should periodically be analyzed for air (nonabsorbable gases or NAGs).
**Refrigerant Recovery Systems**
Efficient recovery of refrigerant from equipment or containers requires evacuation at the end of the recovery cycle. Suction lines to a recovery compressor should be periodically checked for leaks to prevent compressing air into the recovery cylinder during evacuation. In addition, the recovery cylinder pressure should be monitored, and evacuation stopped in the event of a rapid pressure rise, indicating the presence of noncondensable air. The recovery cylinder contents should then be analyzed for NAG, and the recovery system leak checked if air is present. Do not evacuate a refrigeration system that has a major leak.

**Air Monitors and Leak Detection**
Service personnel have used leak detection equipment for years when servicing equipment. Leak detectors exist not only for pinpointing specific leaks, but also for monitoring an entire room on a continual basis. American Society of Heating, Refrigeration and Air Conditioning Engineers (ASHRAE) Standard 15 requires air monitors in refrigeration machinery rooms as defined in the Standard. In conformance with the Standard, an air monitor capable of measuring 0–1500 ppm of HCFC-124 may be required for indoor installations. The user should check ASHRAE Standard 15 to determine what is required for the particular application.

There are several reasons for leak pinpointing or area monitoring, including:
- conservation of refrigerant
- protection of employees
- detection of fugitive or small emissions
- protection of equipment

Leak detectors can be placed into two broad categories: leak pin pointers and area monitors. Before purchasing a monitor or pinpointer, several criteria should be considered, which include sensitivity, detection limits, and selectivity.

**Types of Detectors**
Using selectivity as a criterion, leak detectors can be placed into one of three categories: nonselective, halogen-selective, or compound-specific. In general, as the specificity of the monitor increases, so will the complexity and cost.

A different technology that can be employed to find leaks is by using a dye or fluorescent additive that is placed in the refrigeration system and emitted with the leaking refrigerant and lubricant.

A detailed discussion of leak detection, along with a list of manufacturers of leak detection equipment, can be found in DuPont Bulletin ARTD-27 (H-31753-2).

**Nonselective Detectors**
Nonselective detectors are those that will detect any type of emission or vapor present, regardless of its chemical composition. These detectors are typically quite simple to use, very rugged, inexpensive, and almost always portable. However, their inability to be calibrated, long-term drift, lack of selectivity, and lack of sensitivity limit their use for area monitoring.

Some nonselective detectors designed for use with R-114 may have a lower sensitivity when used with HCFC-124 refrigerant. However, newly designed detectors with good sensitivity for HCFCs are now available. Be sure to consult with the manufacturer before selecting or using a nonselective detector with HCFC-124 refrigerant.

**Halogen-Selective Detectors**
Halogen-selective detectors use a specialized sensor that allows the monitor to detect compounds containing fluorine, chlorine, bromine, and iodine, without interference from other species. The major advantage of such a detector is a reduction in the number of “nuisance alarms”—false alarms caused by the presence of some compound in the area other than the target compound.

These detectors are typically easy to use, feature higher sensitivity than the nonselective detectors (detection limits are typically <5 ppm when used as an area monitor and <0.05 oz/yr [<1.4 g/yr] when used as a leak pinpointer) and are very durable. In addition, due to the partial specificity of the detector, these instruments can be easily calibrated.
**Compound-Specific Detectors**
The most complex detectors, which are also the most expensive, are compound-specific detectors. These units are typically capable of detecting the presence of a single compound, without interference from other compounds. In an area where different refrigerant mixtures are used, these detectors may offer more specificity than is needed for normal leak management. Discuss these issues with the equipment manufacturer before making a purchase decision.

**Storage and Handling**

**Shipping in the United States**
HCFC-124 is a liquefied compressed gas. According to the U.S. Department of Transportation (DOT), a non-flammable compressed gas is defined as a nonflammable material having an absolute pressure greater than 40 psia at 21°C (70°F) and/or an absolute pressure greater than 104 psia at 54°C (130°F).

The appropriate DOT designations are as follows:

<table>
<thead>
<tr>
<th>DOT Proper Shipping Name</th>
<th>LIQUEFIED GAS N.O.S. (CHLOROTETRAFLUOROETHANE)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Hazard Class</td>
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<tr>
<td>UN Number</td>
<td>1021</td>
</tr>
<tr>
<td>DOT/IMO Labels</td>
<td>NONFLAMMABLE GAS</td>
</tr>
</tbody>
</table>

**Handling Precautions for HCFC-124 Shipping Containers**
The following rules for handling HCFC-124 containers are strongly recommended:

- Never refill disposable cylinders with anything. The shipment of refilled disposable cylinders is prohibited by DOT regulations.
- Never refill returnable cylinders without DuPont consent. DOT regulations forbid transportation of returnable cylinders refilled without DuPont authorization.
- Never use a lifting magnet or sling (rope or chain) when handling containers. A crane may be used when a safe cradle or platform is used to hold the container.
- Never use containers as rollers, supports, or for any purpose other than originally intended.
- Protect containers from any object that will result in a cut or other abrasion in the surface of the metal.
- Never tamper with the safety devices in the valves or containers.
- Never attempt to repair or alter containers or valves.
- Never force connections that do not fit. Make sure the threads on the regulators or other auxiliary equipment are the same as those on the container valve outlets.
- Keep valves tightly closed and valve caps and hoods in place when the containers are not in use.
- Store containers under a roof to protect them from weather extremes.
- Use a vapor recovery system to collect vapors from lines after unloading.

**Recovery, Recycle, Reclamation, and Disposal**
Responsible use of HCFC-124 requires that the product be recovered for reuse or disposal whenever possible. DuPont purchases used refrigerant for reclamation through its distributor networks in the United States, Canada, and Europe. In the United States, Suva® 124 refrigerant is accepted as part of this program. Recovery and reuse of refrigerant makes sense from an environmental and economic standpoint. In addition, the U.S. Clean Air Act prohibits known venting of CFC, HCFC, and HFC refrigerants during the maintenance, servicing, or disposal of refrigeration equipment.
**Recovery**

Recovery refers to the removal of refrigerant from equipment and collection in an appropriate container. As defined by the Air Conditioning and Refrigeration Institute (ARI), recovery does not involve processing or analysis of the refrigerants. HCFC-124 may be recovered from refrigeration equipment using permanent on-site equipment or many of the portable recovery devices now available in the marketplace. The portable devices contain a small compressor and some have an air-cooled condenser. In many, the recovery cylinder serves as the condenser. They may be used for vapor (and in some cases, liquid) recovery. At the end of the recovery cycle, the system is evacuated thoroughly to remove vapors. In the United States, the Environmental Protection Agency (EPA) sets standards for recovery equipment. Before purchasing a specific recovery unit, check with the manufacturer to be sure that it contains proper materials of construction and lubricant for the refrigerants you intend to recover.

**Recycle**

Refrigerant recycle refers to reducing the contaminant levels in used refrigerants by passing the refrigerants through devices that separate out or reduce the amount of lubricant, water, acidity, and particulates. Recycle is usually a field or shop procedure with no analytical testing of refrigerant. HCFC-124 may be recycled using many of the devices now on the market. In the United States, the EPA sets standards for these devices. Recycle is already standard practice in many areas of the commercial refrigeration industry. Consult with the manufacturer before specifying a recycle device for any refrigerant.

**Reclamation**

Reclamation refers to the reprocessing of used refrigerant to new product specifications. Quality of the reclaimed product is verified by chemical analysis. In the United States, Suva® 124 refrigerant is included in DuPont’s refrigerant reclamation program. Contact DuPont or one of our authorized distributors for further information.

Reclamation offers advantages over on-site refrigerant recycling procedures, because recycling systems cannot guarantee complete removal of all contaminants. Putting refrigerants that do not meet new product specifications into expensive equipment may cause damage.

**Disposal**

Disposal refers to the destruction of used refrigerant. Disposal may be necessary when the refrigerant has become badly contaminated with other products and no longer meets the acceptance specifications of reclaimers. Although DuPont does not accept severely contaminated refrigerant for disposal, licensed waste disposal firms are available. Be sure to check the qualifications of any firm before sending them used refrigerants.
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